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3.7 Define conductivity and molar conductivity for the solution of an electrolyte. Discuss their variation with concentration. Sol. Conductivity of a solution is defined as the conductance of a solution of 1 cm length and having 1 sq. cm as the area of cross-section. Molar conductivity of a solution at a dilution ( $V$ ) is the conductance of all the ions produced from one mole of the electrolyte ...

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